



## Iodine's Role in Public Health: Review

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### Abstract

Iodine is one of the essential micronutrients in the formation of thyroid hormones in the body, and its deficiency causes pathological problems due to insufficient production of this hormone. The aim of this article is to review important topics related to iodine deficiency and its impact on public health. Iodine deficiency is the main cause of endemic goiter, and eating some foodstuffs in quantities exceeding the permissible amounts may interfere with the absorption of iodine in the thyroid gland. Hypothyroidism caused by iodine deficiency early in life can irreversibly impair neuromotor development, leading to poor cognition and growth in children. The prevalence of iodine deficiency disorders worldwide occurs mostly in low- and middle-income countries. A healthy, balanced diet is the ideal solution to provide sufficient iodine. The most common foods that contain iodine are dairy products, seafood, and iodized salt.

**Keywords:** Iodine deficiency, pathological problems, hypothyroidism, public health.

### 1. Introduction

Micronutrients are essential for the body to perform its vital functions properly. A deficiency of these rare nutritional elements, or the occurrence of an overlapping of the elements and the loss of their nutritional value, can incite disturbances in the body's nutritional metabolism, ruining the body's health [1,2]. As any of these nutrients is deficient, disease problems appear, worsen, and ultimately lead to death [3]. Iodine is a rare and essential element in the synthesis of thyroid hormones, which are necessary for the metabolism of all body cells [4]. In addition to its crucial role in the neurological development of the fetus and child, therefore adequate iodine nutrition is imperative for all mammals [5,6]. Iodine is often found in very small concentrations in the body fluids and tissues of animals and humans. It has been proven that five grams of iodine is sufficient for a person's needs for seventy years of life [7,8]. Dietary iodine is incorporated by intestinal epithelial cells into the blood and then into the thyroid gland, where 80% of the body's total iodine is stored and used for the synthesis of the thyroid hormones 3,3,5-triiodothyronine (T<sub>3</sub>) and thyroxin (T<sub>4</sub>). A healthy adult body contains 15-20 mg of iodine, approximately 70-80% of which is found in the thyroid gland [9-11]. Iodine deficiency is the main cause of thyroid disorders and a host of other health problems; it also causes lifelong brain damage, mental retardation, infertility, and lower infant survival rates [12,13]. This review presents the topics of the element iodine and its cycle in nature, the sources of obtaining it, the body's needs, and the disorders resulting from its deficiency, and provide better information about that.

## 2. Iodine element

Iodine is a non-metallic chemical element located in group seventeen and has the symbol (I), atomic number 53, and atomic mass 126.9 g/mol, nearly black solid at standard conditions ( figure 1). It is the heaviest of the stable halogens [14,15]. It is mono-isotopic and has only one stable isotope, I-127. It melts to form a deep violet liquid at 114°C, and boils to a foul-smelling violet gas at 184°C. It is non-flammable and toxic, and does not combine with oxygen directly. In nature, it does not exist in its free state and forms diatomic molecules [16,17]. Iodine and iodate are the most common inorganic forms of iodine in the environment [18]. In 1811, it was discovered by French chemist Bernard Courtois when he accidentally added concentrated sulfuric acid to Fucus vesiculosus, seaweed used to treat goiter. A few years later, specifically in 1813, iodine was named by the French scientist Joseph Louis Gay-Lussac as a new element, and its name is derived from the Greek word iodes, which means “violet color” [19]. Iodine is primarily used in medicine to disinfect external wounds, especially recent wounds, due to its high ability to kill germs and other microorganisms that may be harmful [20]. It is also used to improve the functioning of a sluggish thyroid gland, while its solution is used to treat swelling of the thyroid gland [21]. For several years, iodine has been used in the manufacture of some anti-radiation drugs because it is the only element capable of preventing the accumulation of radioactive iodine and iodide salts in the thyroid gland [22]. In addition, silver iodide is used in photography, as well as in the manufacture of dyes [23].

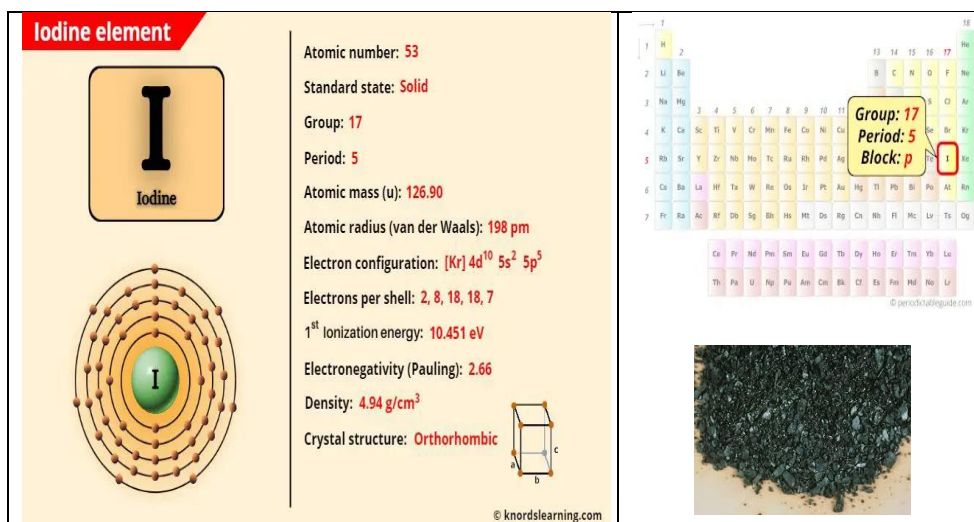


Figure 1: Iodine (I) element.

## 3. Iodine in nature

Iodine (as iodide) is unevenly present in the Earth's environment and can be found in many geological formations. Although iodine is concentrated in seawater, the biosphere, and the atmosphere, but seawater represents by far the biggest reservoir of iodine[24]. Topsoils have been depleted of iodide due to leaching from glaciation, flooding and erosion in many areas, so most of

the iodide in the natural environment is concentrated in the oceans. In natural iodine cycle (figure2), marine iodide ions are oxidized to elemental iodine, which evaporate and volatilize through the atmosphere and when rain falls, it returns to the soil [25,26]. Thus, many cases of iodine deficiency disorders have been found to occur in areas of high mountain ranges, rain shadows, and central continental regions. In general, little iodine in the secondary environment is derived from weathering of the lithosphere, and most of the iodine is derived from volatilization from the oceans and later transported to land [27,28]. However, iodine cycles are often slow and incomplete, leaving soil and drinking water depleted of iodine. Iodine-deficient soils are common in mountainous areas and areas where flooding is common [29]. Crops grown in these soils will therefore be low in iodine and people and animals who consume food grown in these soils will become iodine deficient [30]. On the other hand, iodine was found to be concentrated in sulfur-containing minerals, which led to its classification as a chalcophilic element. The iodine content of the various rock-forming minerals is homogeneous and fairly low, with some enrichment found only in the chlorine-containing minerals sodalite and iodialite. Because of the large ionic radius of the iodide ion (220  $\mu\text{m}$ ), iodine does not enter the crystal lattices of most rock-forming minerals [31]. Sedimentary rocks have a wider range of iodine content with clay-rich rocks or clay rocks being more enriched than sand-rich sandstones [32]. Soils in coastal areas are higher in iodine than mountain soils (0.24  $\mu\text{g}$  iodine/g of soil), since they evaporate from the sea and deposit in the surrounding areas [33].

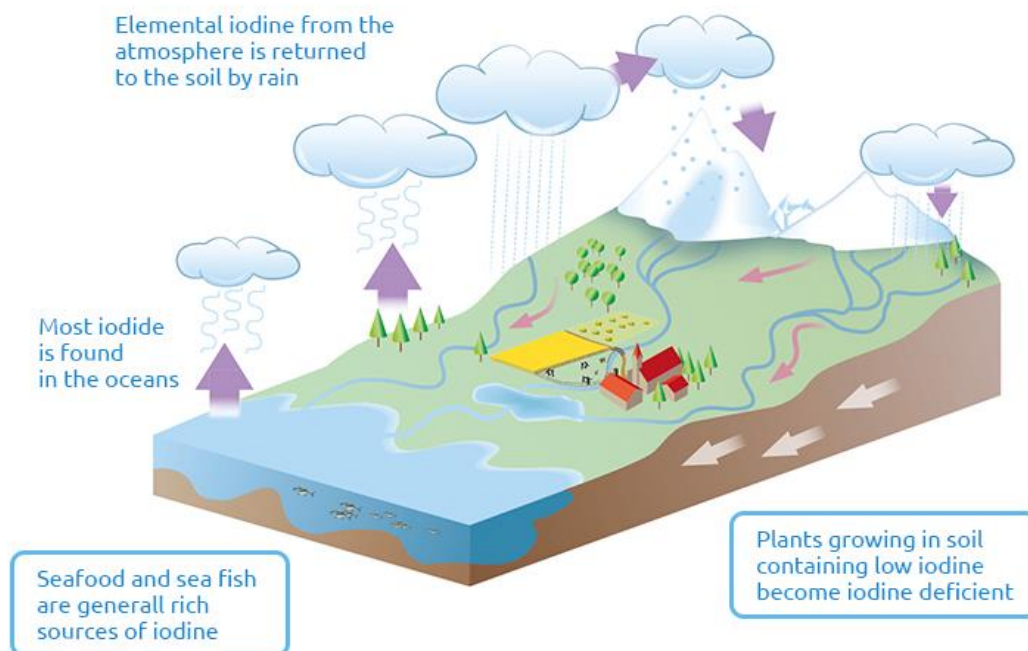


Figure 2: The natural iodine cycle.



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## 4. Food sources of iodine

The amount of iodine found in natural foods varies greatly depending on the soil in which plants and animals grow. Foods that contribute to high iodine intake are usually (figure 3) bread, dairy products, iodized salt, eggs, fish and seaweed [34]. Seaweed has a great ability to store iodine from the sea, so it is considered one of the foods richest in iodine [35]. Although lettuce and spinach contain less iodine than seaweed, they are a valuable source of iodine for humans. Serving size should be taken into consideration when evaluating the actual contribution of foods naturally high in iodine to dietary intake [36]. On the other hand, fortified foods can contribute more iodine in smaller portions, as observed with bread samples made with iodine dough improvers. Changing dietary patterns and food processing techniques in the modern era may contribute to increased iodine deficiency [37]. Iodine-containing compounds used in fertilizers and irrigation affect plants that are consumed by livestock for feed [38]. Reducing the use of iodophors as sterilizing agents in milk processing may affect the iodine content of dairy products [39]. As well as increased avoidance of dairy products, along with the emergence of plant-based alternative milk sources, contributes to a decrease in iodine status [40]. Also, reduced use of dietary salt for health reasons that consider salt a risk factor for blood hypertension is a contributing factor to reduced intake of iodized salt [41]. However, up to 20% of the iodine in salt may be lost during processing and another 20% during food preparation. To overcome poor iodine intake, strategies have been developed, including fortification programs aimed at incorporating iodine into common foods such as milk, bread, water and salt. According to the World Health Organization (WHO), the most successful strategy is the Universal Salt Iodization (USI) program. Since salt is consumed fairly by most of the population throughout the year [42-44].

Food Item	Amount	Approximate (mcg) per Serving	Percent of Daily Value
 Banana	Medium one	3	2
 Canned Corn	Half cup	14	9
 Plain Yogurt low fat	One cup	75	50
 Iodized Salt	One and half gram	71	47
 Boiled Macaroni	One cup	27	18
 Raisin Bran Cereal	One cup	11	7
 Seaweed	One gram	16 to 2984	11 to 1989
 Shrimp	Three ounces	35	23
 Reduced Fat Milk	One cup	56	37
 Egg	One piece	24	16
 Fruit Cocktail	Half cup	42	28
 Canned Tuna Fish	Three ounces	17	11
 Chocolate Ice-cream	Half cup	30	20
 Cheddar Cheese	One ounce	12	8
 Fish Sticks	Three ounces	54	36
 Baked Cod	Three ounces	99	66

Figure 3: Food sources of iodine and their value.

## 5. Iodine nanoparticles (INPs) applications

Nanotechnology has revolutionized the field of medicine, just like many other fields [44-46]. Nanoparticles (NPs) have unique features due to their small size (1-100 nm) and their larger



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surface area to volume ratio[47,48], in addition to their physicochemical properties,[49] which allowed them to be exploited in various medical applications, as a drug delivery system[50], biosensors[51], and diagnosis and treatment of malignant tumors [52,53]. In the current era, researchers are interested in the applications of these particles [54-57], especially iodine nanoparticles (INPs), as therapeutic agents in the field of cancer diagnosis and treatment [58]. Among the strategies used in cancer treatment is radiotherapy using radiopharmaceutical agents, especially radioisotopes of iodine, and there are investigations conducted to develop the new structure of INPs [59]. It is worth noting that iodine nanoparticles can be used in nuclear medicine imaging and may be effective with high voltage (MV) photons in treating cancer, but it requires evaluation using different cancer cells. Combining iodine with other therapeutic techniques has increased the effectiveness of diagnosing and treating malignant tumors. The use of various radioisotope-labeled nanoparticles, especially iodine-125, has been reported in several research works as a nuclear imaging agent for in vivo imaging and bio-distribution analysis. Recently, INPs have attracted increasing attention due to their potential applications in both imaging and therapy [60]. Previously, during the past two decades; nano-based contrast agents have revolutionized preclinical imaging science, facilitating early, rapid and effective cancer detection. Imaging techniques can also be used in conjunction with nanotechnology-based contrast agents to investigate drug interactions at the preclinical stage and at the cellular level [61]. Iodine nanoparticles have been used by intravenous injection into human TNBC tumors growing in the brains of thymic nude mice. High uptake in the vicinity of the tumor was found by micro-CT, reaching 4.5% of iodine by weight. A better treatment for human brain metastases may be achieved by combining INP-RT with synergistic chemotherapy and other treatments, such as immunotherapy, oncolytic virus therapy, using optimized irradiation equipment [62].

## 6. Iodine deficiency disorders

Iodine deficiency has devastating effects on the growth and development of the body and causes a group of diseases called iodine deficiency disorders (IDD), which are considered one of the most important human diseases and have the most impact on public health [63]. These disorders result from insufficient production of thyroid hormone due to a lack of sufficient iodine (Figure 4). It is the most common cause of goiter, and it is estimated that it affects approximately 2 billion people in the world's population, and the incidence of goiter depends on the degree of iodine deficiency. In general, it was more common in females [64,65]. Hypothyroidism, one of the symptoms of severe iodine deficiency, in Southeast Asia and the Eastern Mediterranean, studies have shown that iodine deficiency disorder exists in 36% and 43% of the total population, respectively. Individuals with mild deficiency are at risk for goiter, the formation of benign thyroid nodules with or without goiter, and the formation of endemic goiter [66]. As iodine intake decreases, the secretion of pituitary thyroid-stimulating hormone increases to maximize the uptake of circulating iodine by the thyroid gland, and thus thyroid-stimulating hormone stimulates thyroid enlargement and enlargement [67]. In adults, severe iodine deficiency manifests as mental disability, and decreased fertility. Breast cancer in women is associated with low levels of iodine in the body, and the use of iodine improves the adaptations that occur in some women's breasts [68]. In children, goiter, impaired intellectual and physical development, deafness and cretinism can occur.

Cretinism is the most serious neurological damage resulting from hypothyroidism in the fetus and is characterized by gross mental retardation with varying degrees of short stature, deafness, mutism, and spasticity [69]. Multi-nodular goiter can be classified as hypothyroid and toxic depending on the clinical presentation and molecular pathology. Mutations that facilitate the growth of thyroid cells can lead to thyroid cancer in multi-nodular goiter [70]. One of the consequences of goiter is obstruction of the trachea and esophagus, which incites damage to the recurrent laryngeal nerves [71]. Severe iodine deficiency during pregnancy increases the risk of stillbirth, miscarriage, and birth defects [72]. In general, iodine deficiency leads to adverse health effects at all ages, and reduces learning ability and productivity [73].

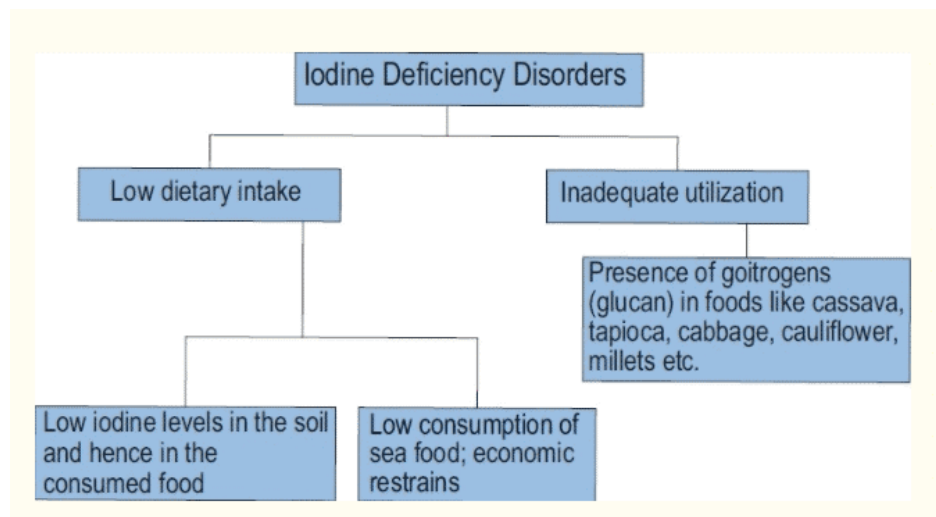


Figure 4: Causes of iodine deficiency.

## 7. Conclusions and recommendations

Iodine is necessary for the thyroid gland to function properly and avoid its enlargement, and thus plays a pivotal role in cognitive and physical maturity. Iodine deficiency disorders can lead to mental impairment in children. Cretinism is one of the most severe forms of iodine deficiency disorder, which poses a major threat to the health and development of the world's population. Therefore, this article recommends improving diets to prevent iodine deficiency by diversifying food sources, especially eating seafood, which is effective in reducing iodine deficiency. As well as iodized salt and adding iodine to animal and poultry feed to contribute to increasing the amount of iodine present in milk and eggs, and thus contributing to raising dietary iodine levels.

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